

Effects of phosphorus addition on N₂O emission in an *Acacia mangium* plantation with and without root exclusion

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Abstract

Effects of phosphorus (P) fertilization on N₂O emissions in an *Acacia mangium* plantation were compared among C-plot (control), T-plot (trenched), P-plot (P added), and TP-plot (both trenched and P added) for 106 days in the wet season. N₂O emissions from P-plot were significantly smaller than that of C-plot, and are attributed to accelerated root absorption of N. In contrast, N₂O emissions from TP-plot were not significantly smaller and rather seemed to be larger than that of T-plot. We concluded that P application suppress N₂O emissions from *Acacia mangium* plantations, possibly because P addition accelerates root absorption of N.

Key Words

Nitrous oxide, humid Tropics, P limitation, fast-growing wood plantation.

Introduction

It is well known that nitrous oxide (N₂O) is an important greenhouse gas following CO₂ and CH₄ (IPCC, 2007). Soils are major sources for N₂O, which is a byproduct or intermediate product of microbial nitrification and denitrification (Bremer *et al.* 1997; Davidson *et al.* 2000; Wrage *et al.* 2001).

Tropical forests are important sources of N₂O (Matson and Vitousek 1990). Several studies have demonstrated that *Acacia mangium* was a non-negligible emission source of N₂O (Dick *et al.* 2006; Arai *et al.* 2008; Konda *et al.* 2008) because of *Acacia*'s high N-fixing ability as a leguminous tree.

Hall and Matson (1999) suggested that poor soil P availability, which is common in tropics (Vitousek *et al.* 1993, Cleveland *et al.* 2002), causes higher emissions of N₂O by demonstrating that nitrogen (N) fertilizer inputs to the P-limited ecosystems generated 10-100 times greater N₂O fluxes than the same treatment to the N-limited ecosystems. Their report suggests that the removal of P limitation may lead to suppression of N₂O and NO₂ loss. Here we tested the hypothesis that the removal of P limitation suppresses N₂O emission from *Acacia mangium* plantations. We used trench method to investigate the effects of P fertilization on N₂O emission both in vegetation-existing condition and non-vegetation-existing condition in an *Acacia mangium* plantation.

Materials and methods

Plot setting

The experiment site is located at an *Acacia mangium* plantation in South Sumatra Province, Indonesia (3°47.394' E, 103°55.236' S). This region is humid tropical rain forest with an annual precipitation of 2000~3000 mm and a mean annual temperature of 27.3°C (Hardjono *et al.* 2005). The period from April to September is a relatively dry season and from October to May is wetter season, although there is no clear distinction. The soils in the area are Acrisols with Tertiary sedimentary rock as the parent material.

A 6-year-old *Acacia mangium* stand, where trees were planted 3 m interval, was selected for the experiment. The experimental design included four treatments, C-plot (control), T-plot (trenched), P-plot (P added), and TP-plot (both trenched and P added). Plots of 9 m×9 m were established randomly in 6 replications for every treatment. In T-plots and TP-plots, we established 1 m×1 m subplots and made vertical cuts along the boundaries down to 0.5 m below the ground surface (approximately the bottom of the root zone) at the beginning of September 2008. Pieces of polyethylene board were inserted into the vertical cuts to inhibit root re-growth. Soils were refilled into the trench to minimize disturbance as little as possible. Seedlings and herbaceous vegetation in T-plots and TP-plots were removed by hand when necessary. At the beginning of November, TSP (Triple Super Phosphate) was applied to each P-plot and TP-plot by hand at the rate of 200 kg/P/ha/yr.

Gas flux measurement

From the end of November to the beginning of March, N₂O emissions were measured 6 times (3-day before and 4-, 7-, 15-, 29-, and 106-days after P addition) with a static chamber method (Ishizuka *et al.* 2002), using PVC chambers (20 cm diameter, 15 cm height). We chose this period because N₂O emissions were reported to be larger in the wetter season (Arai *et al.* 2008). Forty-ml gas samples at 0, 15, and 30 min were taken into 30-ml glass vacuum vials with butyl rubber lids. N₂O concentration was analyzed using a gas chromatograph (GC-14B, SHIMADZU, Kyoto, Japan) equipped with an electron capture detector. We calculated N₂O emission rates using a linear regression slope, using data at 0, 15, and 30 min, because the increase in gas concentration appeared linear. The cumulative gas emissions were estimated using the linear trapezoidal method.

Soil sampling and analysis

At each gas sampling day, soils were also sampled from 0-5 cm depth. We collected a single soil core within 1 m radius of each chamber using 100 mL cylinder. Soil water content was immediately measured using a portion of the sample. WFPS was calculated as follows:

$$\text{WFPS} = (\text{Gravimetric water content}) \times (\text{Bulk density/Porosity}) \times 100, \quad (1)$$

where Porosity = 1 – (Bulk density/Particle density). Soil pH (H₂O) was measured using moist soils. Ten g soil and distilled water were mixed to 1 : 2.5 ratio, and pH was measured after shaking 30 min.

Data analysis

The level of significance was examined by ANOVA followed by Tukey's multiple comparison tests after confirming the normality of data by the Kolmogorov-Smirnov test. Each statistical analysis was performed using SPSS version 10.0 (SPSS Inc., Chicago, USA).

Results and discussion

The values of pH did not change during experiment period. N₂O emissions from P-plot were significantly smaller than that of C-plot (Figure 1). In contrast, N₂O emissions from TP-plot were not significantly smaller and rather seemed to be larger than that of T-plot (Figure1), showing that P application suppressed N₂O emissions only if vegetation existed. Figure 2 shows the relationship between N₂O emission rate and WFPS at 4-, 7-, 15-, 29-, and 106-days after the start of experiment. N₂O emissions seemed to be smaller in P-plot than C-plot in a similar WFPS condition, while there were no such trends for TP-plot and T-plot.

Earlier studies suggest that P addition to P-limited soils reduces N₂O emissions by removing the P limitation on microbial N immobilization and decreasing N for nitrification and denitrification processes.

Sundareshwar *et al.* (2003) showed that P addition to swampy ecosystem reduced the denitrification rate and N₂O emissions. In our study, however, such a mechanism did not operate, for P addition to T-plot did not suppress N₂O emissions significantly.

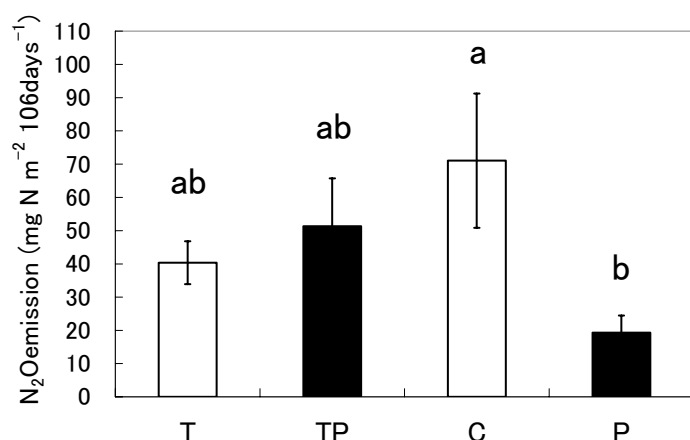


Figure 1. The cumulative N₂O emissions from each subplot. The same letters are not significantly different using ANOVA followed by Tukey's multiple comparison tests ($P < 0.05$).

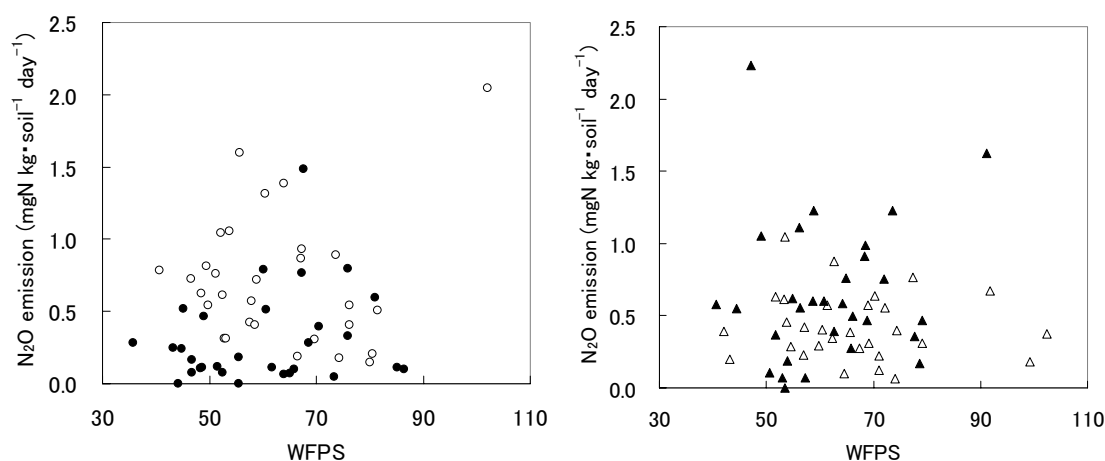


Figure 2. Relationship between N₂O flux and WFPS at 4, 7, 15, 29, and 106 days after the start of experiment. Open and closed circles represent C-plots and P-plots, respectively, and open and closed triangles represent T-plots and TP-plots, respectively.

Meanwhile, P addition significantly decreased N₂O emissions from the plot without trenching, probably because P addition accelerated root absorption of N. Vegetation is presumably the main competitor of nitrifying and denitrifying microbes because plant growth is dependent on N availability as well as N₂O emission (Aerts *et al.* 1995). Although it has been recognized in various forest ecosystems that microorganisms are stronger competitors for inorganic N than plants, (Johanson, 1992; Kaye and Hart, 1997), some studies have that particular plant species to be the stronger competitor (Silvan *et al.* 2005). Probably *Acacia mangium* is also stronger competitor for inorganic N than nitrifying and denitrifying microbes, thereby the P addition released P limitation of *Acacia mangium* and increased N uptake and suppressed N₂O emissions.

Conclusion

This study showed that P application should suppress N₂O emissions from the *Acacia mangium* plantation, possibly because P addition accelerates root absorption of N. P application to the *Acacia mangium* plantation might be an effective global warming mitigation option by suppressing N₂O emissions

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